

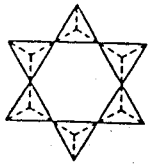

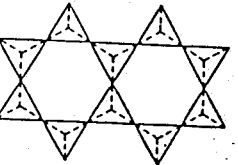
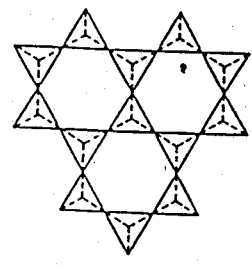
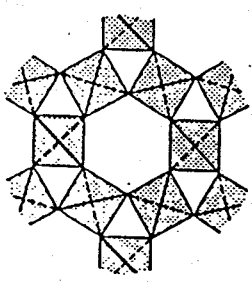


Table 1-2. Common nonsilicate minerals in soils.

Mineral class	Mineral	Chemical formula
Halides	Halite	NaCl
Sulfates	Gypsum	CaSO ₄ ·2H ₂ O
	Jarosite	KFe ₃ (SO ₄) ₂ (OH) ₆
Sulfides	Pyrite	FeS ₂
Carbonates	Calcite	CaCO ₃
	Dolomite	CaMg(CO ₃) ₂
	Nahcolite	NaHCO ₃
	Trona	Na ₂ CO ₃ ·NaHCO ₃ ·2H ₂ O
	Soda	Na ₂ CO ₃ ·10H ₂ O
Oxides and Hydroxides		
Aluminum	Gibbsite	Al(OH) ₃
	Nordstrandite	Al(OH) ₃
	Boehmite	AlOOH
	Diaspore	AlOOH
	Corundum	Al ₂ O ₃
Iron	Hematite	Fe ₂ O ₃
	Goethite	FeOOH
	Lepidocrocite	FeOOH
	Maghemite	Fe ₂ O ₃
	Ferrihydrite	Fe ₅ O ₇ (OH)·4H ₂ O
Manganese	Magnetite	Fe ₃ O ₄
	Lithiophorite	(Al,Li)MnO ₂ (OH) ₂
	Birnessite	variable
	Hollandite	Ba ₂ Mn ₈ O ₁₆
	Pyrolusite	MnO ₂
	Todorokite	variable
Titanium	Manganite	MnOOH
	Rutile	TiO ₂
	Anatase	TiO ₂
	Ilmenite	FeTiO ₃

Classification of silicates

Silicate class	Arrangement of tetrahedra ^a	Structural group	T:O ^b ratio	Examples	Structural formula ^c
Nesosilicates		Independent Tetrahedra	1:4	Olivine Zircon Garnet Spinel Epidote Hemimorphite	$(Mg,Fe)_2SiO_4$ $ZrSiO_4$ $R_2^+R_3^+(SiO_4)_3$ $CaTi(SiO_4)_2$ $Ca_2(Al,Fe)_2(SiO_4)_2(OH)_2$ $Zn_4(Si_2O_7)(OH)_2H_2O$
Sorosilicates		Double Tetrahedra	2:7		
Cyclosilicates		Hexagonal rings of Tetrahedra	1:3	Tourmaline	$Na(Mg,Fe)_3Al_3BO_3Si_3O_{11}$
Inosilicates		Continuous single Tetrahedral chains	1:3	Pyroxenes: Augite Hypersthene	Enstatite $Ca(Mg,Fe,Al)[(Si,Al)_2O_6]$ $(Mg,Fe)_2(Si_2O_6)$
Inosilicates		Continuous double Tetrahedral chains	4:11	Amphibole: Hornblende Actinolite	$Ca_2Na(Mg,Fe)_5(Al,Si)_7O_{22}(OH,F)_2$ $Ca_7(Mg,Fe)_3Si_8O_{22}(OH)_2$
Phyllosilicates		Continuous Tetrahedral Sheets	2:5	Pyrophyllite Talc Micas: Muscovite Illite Biotite Vermiculite Smectite ^e Kaolinite Serpentine	$Al_2Si_4O_{10}(OH)_2$ $Mg_3Si_4O_{10}(OH)_2$ $KAl_2(Si,Al)_4O_{10}(OH)_2$ $K_xAl_3(Si_{4-x}Al_x)_3O_{10}(OH)_2$ $K(Mg,Fe)_3(Si,Al)_3O_{10}(OH)_2$ $(Mg,Fe)_3(Si_{4-x}Al_x)_3O_{10}(OH)_2$ $(Al_{1-x}Mg_x)_2Si_4O_{10}(OH)_2$ $Al_2Si_2O_7(OH)_2$ $Mg_3Si_2O_7(OH)_2$
Tectosilicates		Three-dimensional Tetrahedral framework	1:2	Quartz Opal Feldspars: Orthoclase Albite Anorthite Feldspathoid: Nepheline Zeolite: Analcite	SiO_2 $SiO_2 \cdot nH_2O$ $KAlSi_3O_8$ $NaAlSi_3O_8$ $CaAl_2Si_2O_8$ $NaAlSi_3O_8$ $NaAlSi_3O_8 \cdot H_2O$

Plagioclases, Feldspars:
Anorthite
Bytownite
✓ Labradorite
Albite
✓ Oligoclase
Andesine

β -quartz projected on 0001

^a Tectosilicate (β -quartz) arrangement was adapted from Deer et al. (1963); all other silicate arrangements were adapted from Dennen (1960).
^b T = Tetrahedral cation (Si or Al).
^c Formulas are only approximate in such minerals as tourmaline, augite and hornblende. Interlayer cations and water in vermiculite and smectite are not listed.
^d R¹⁺ = Mg, Fe, Ca, Mn; R²⁺ = Al, Fe, Cr.
^e Formula given is for montmorillonite; beidellite and nontronite have significant Al substituting for Si in the tetrahedral position and Fe²⁺ substituting for Al in the octahedral position, respectively. Interlayer cations are not listed.

From Allen and Fanning, 1983. In Wilding et al. (ed.)

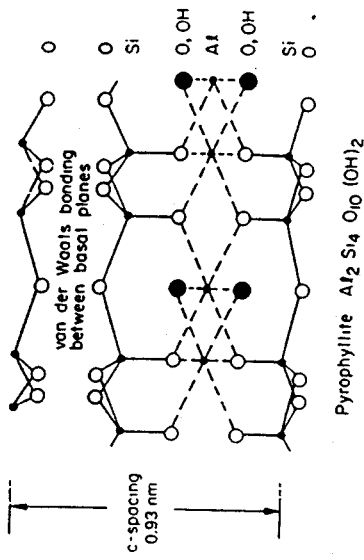


Fig. 4.5 Schematic structure of pyrophyllite. (From F. E. Bear (Ed.), *Chemistry of the Soil*. ACS Monograph Series No. 160, 1969.)

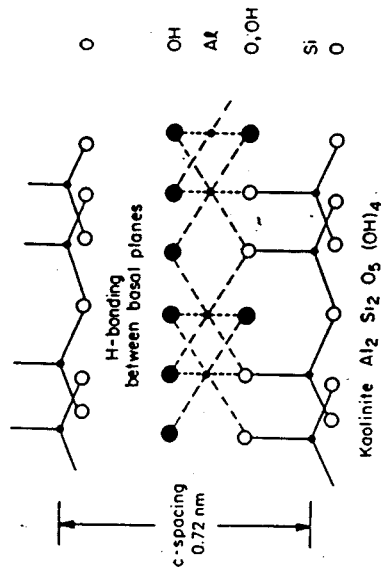


Fig. 4.4 Schematic structure of kaolinite. (From F. E. Bear (Ed.), *Chemistry of the Soil*. ACS Monograph Series No. 160, 1964.)

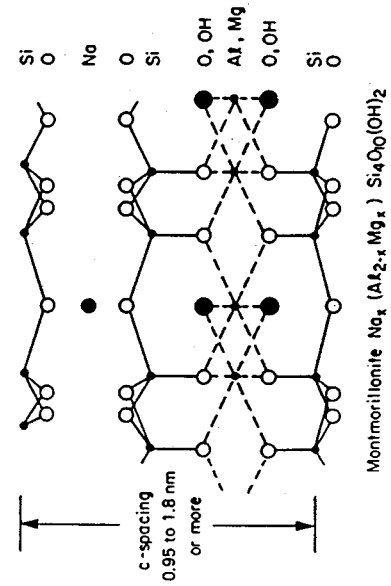


Fig. 4.7 Schematic structure of montmorillonite. (From F. E. Bear (Ed.), *Chemistry of the Soil*. ACS Monograph Series No. 160, 1964.)

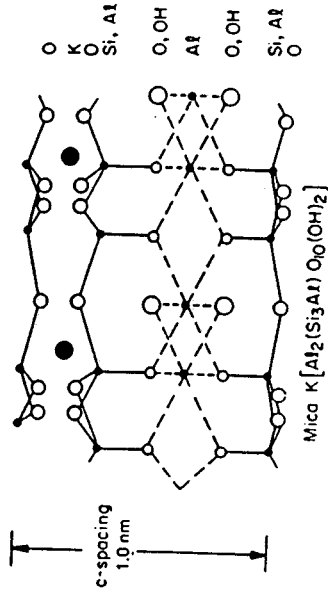


Fig. 4.9 Schematic structure of mica. (From F. E. Bear (Ed.), *Chemistry of the Soil*. ACS Monograph Series No. 160, 1964.)

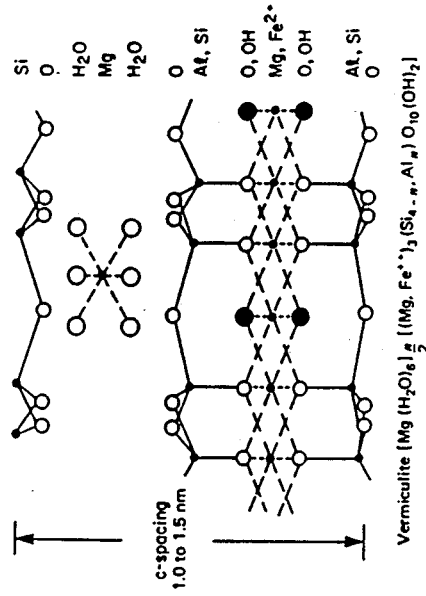


Fig. 4.8 Schematic structure of vermiculite. (From F. E. Bear (Ed.), *Chemistry of the Soil*. ACS Monograph Series No. 160, 1964.)

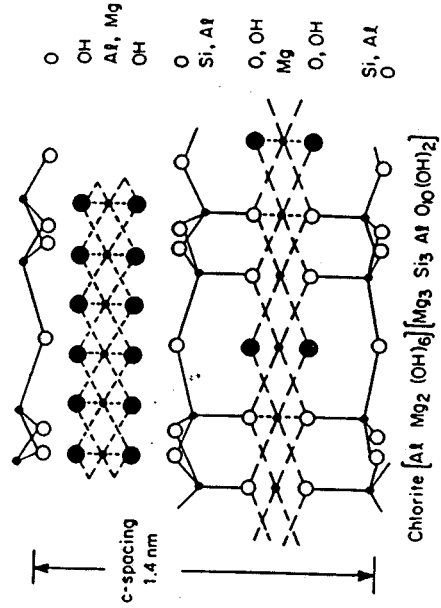


Fig. 4.6 Schematic structure of chlorite. (From F. E. Bear (Ed.), *Chemistry of the*

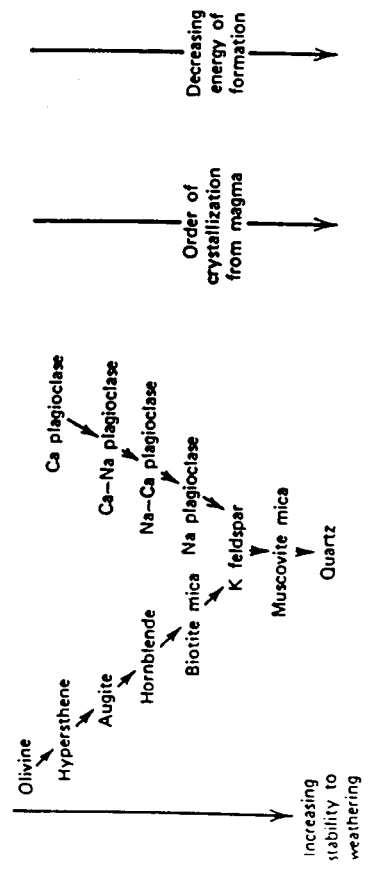


Fig. 3.2 Stability to weathering of some minerals in igneous and metamorphic rocks. (Adapted from S. S. Goldich, *J. Geology*, 46:38 (1938).)

Products of weathering.

Mineral	Residual Products	Material in Solution
Quartz	Quartz grains	Silica
Feldspars	Clay minerals	Silica K ⁺ , Na ⁺ , Ca ²⁺
Amphibole (hornblende)	Clay minerals	Silica Ca ²⁺ , Mg ²⁺
Olivine	Limonite Hematite Limonite Hematite	Silica Mg ²⁺

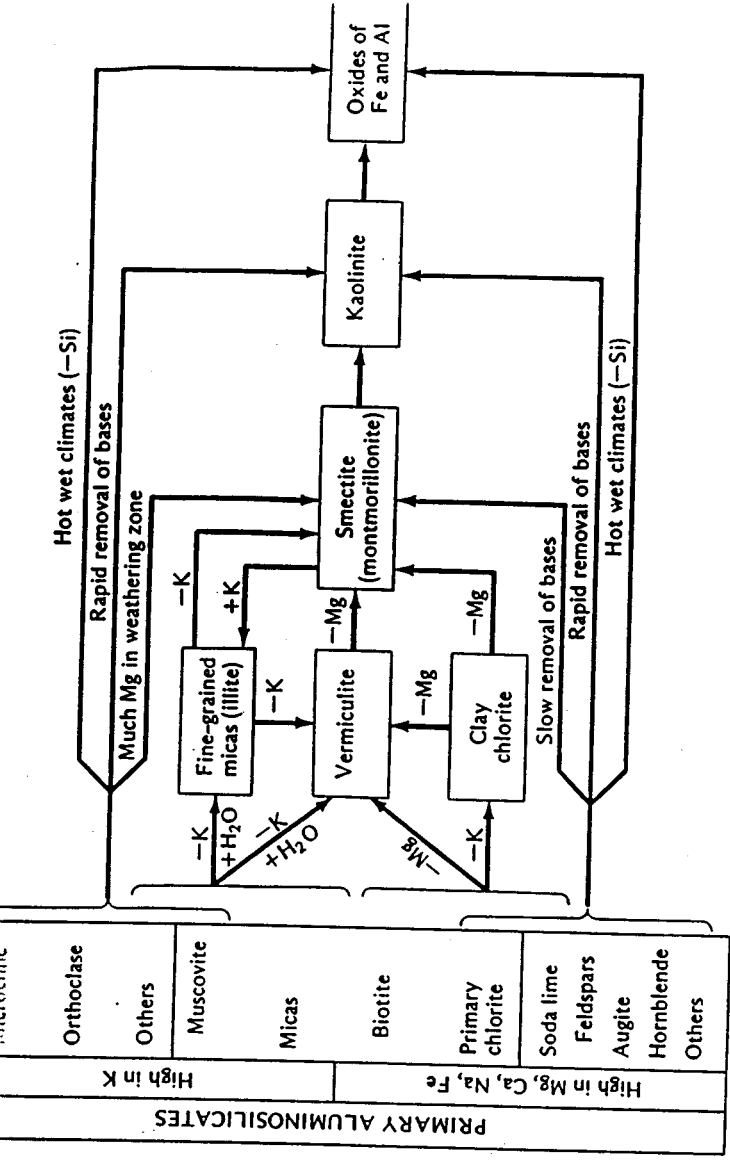
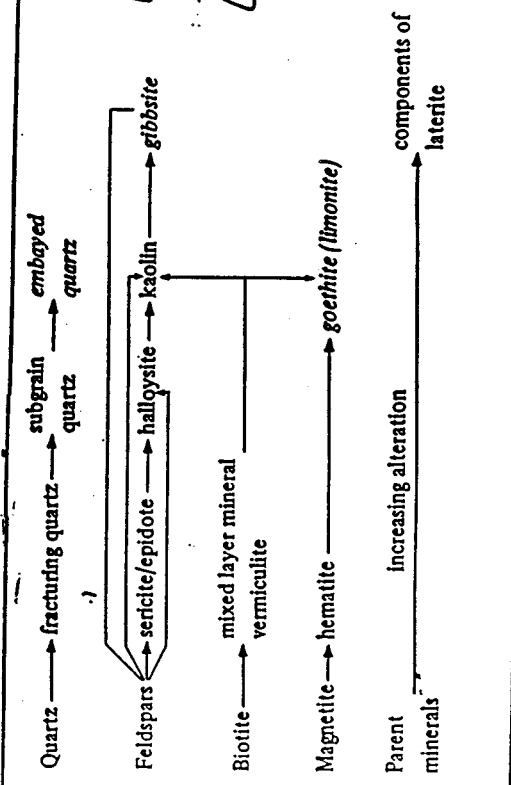
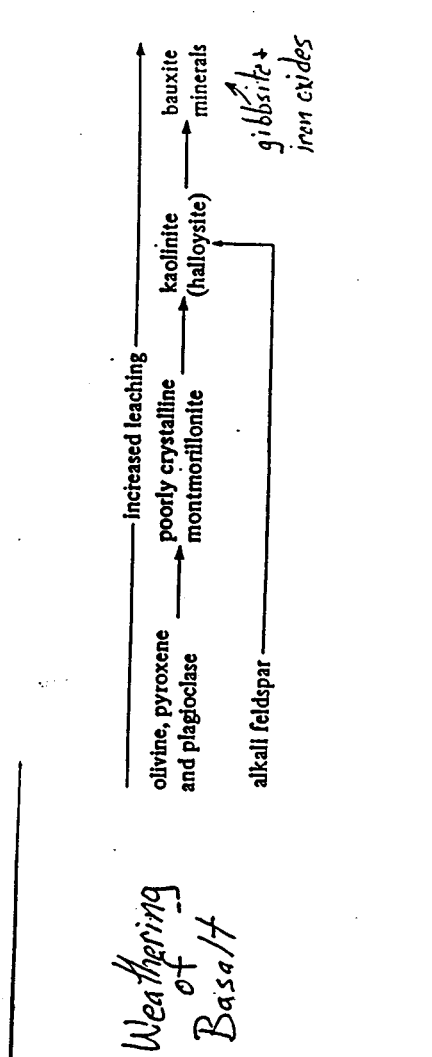
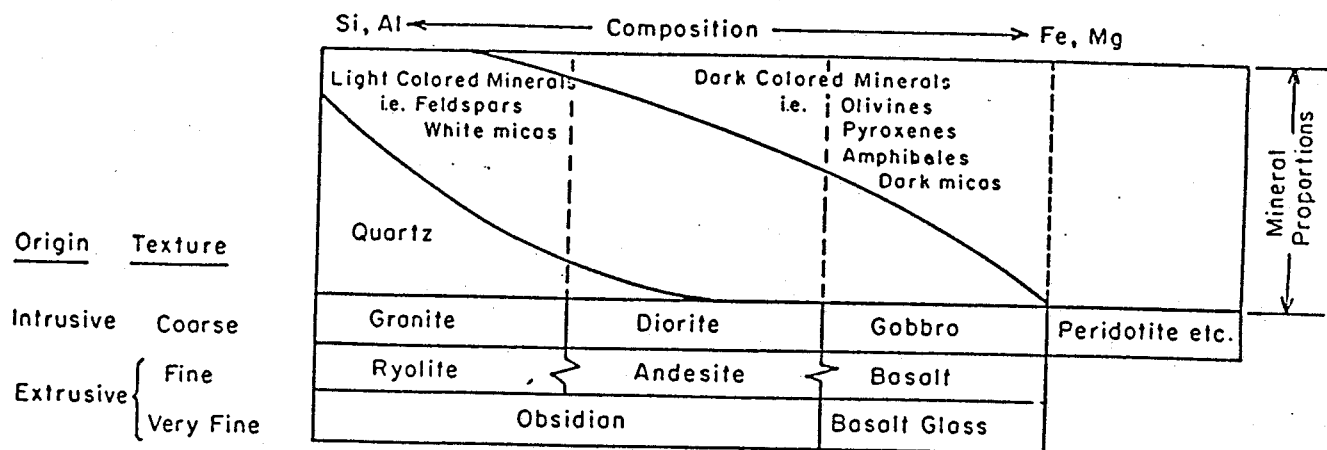


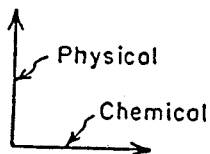
FIGURE 7.10 General conditions for the formation of the various layer silicate clays and oxides of iron and aluminum. Fine-grained micas, chlorite, and vermiculite are formed through rather mild weathering of primary aluminosilicate minerals, whereas kaolinite and oxides of iron and aluminum are products of much more intense weathering. Conditions of intermediate weathering intensity encourage the formation of smectite. In each case silicate clay genesis is accompanied by the removal of soluble elements such as K, Na, Ca, and Mg.



SILICATE MINERAL DISTRIBUTION - IGNEOUS ROCKS



Increasing Susceptibility to Weathering



	CHIEF FELDSPAR ALKALI		ALKALI and SODA-LIME		CHIEF FELDSPAR SODA-LIME			FELDSPARS ABSENT	
	Potash Feldspars: Microcline or Orthoclase	Quartz: 0-5%	Quartz: greater than 5%	Orthoclase-Plagioclase	No quartz	Quartz	No quartz	No quartz = Olivine	No quartz Olivine Other Fe-Mg minerals
Percentage of silica	66-55%	75-65%	75-65%	65-50%	70-62%	65-50%	60-45%	50-30%	
Family or clan	Syenite	Granite	Quartz monzonite	Monzonite	Quartz diorite	Diorite	Gabbro	Peridotite	
PLUTONIC	Characteristic texture	Syenite	Granite Alaskite	Quartz monzonite	Monzonite	Quartz diorite	Diorite	Gabbro Anorthosite Norite	Peridotite Dunite
	1 Granitic (megacrystalline)	Syenite	Granite Alaskite	Quartz monzonite	Monzonite	Quartz diorite	Diorite	Gabbro Anorthosite Norite	Peridotite Dunite
HYBAYSAL	2 Porphyritic Ground mass	Syenite porphyry (a mega-)							
	a megacrystalline b microcrystalline Phenocrysts		Rhyolite porphyry (b micro-)	Porphyries					
VOLCANIC	3 Felsitic	Trachyte	Rhyolite	Quartz Latite	Latite	Dacite	Andesite (less than 50% dark minerals)	Basalt (greater than 50% dark minerals)	Various terms in wide use

TABLE 3.1
Common igneous rocks.

	Granitic	Andesitic	Basaltic	Ultramafic
Intrusive Extrusive	Granite Rhyolite	Diorite Andesite	Gabbro Basalt	Peridotite
Mineral Composition	Quartz Potassium feldspar Sodium feldspar	Amphibole Intermediate plagioclase feldspar Biotite	Calcium feldspar Pyroxene	Olivine Pyroxene
Minor Mineral Constituents	Muscovite Biotite Amphibole	Pyroxene	Olivine Amphibole	Calcium feldspar

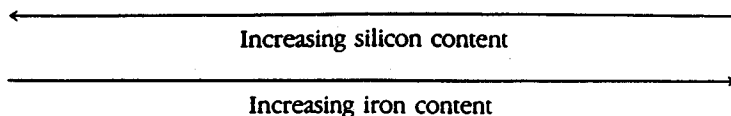


TABLE 4.1
Variations in properties among
magmas of differing
compositions.

Property	Basaltic	Andesitic	Granitic
Silica content	Least (~ 50)	Intermediate (~ 60)	Most (~ 70)
Viscosity	Least	Intermediate	Highest
Tendency to form lavas	Highest	Intermediate	Least
Tendency to form pyroclastics	Least	Intermediate	Highest
Density	Highest	Intermediate	Lowest
Melting point	Highest	Intermediate	Lowest
Typical minerals	Ca feldspar Pyroxene Olivine	Na feldspar Amphibole Pyroxene Mica	K feldspar Quartz Mica Amphibole

SOURCE: Modified from Peter Francis.

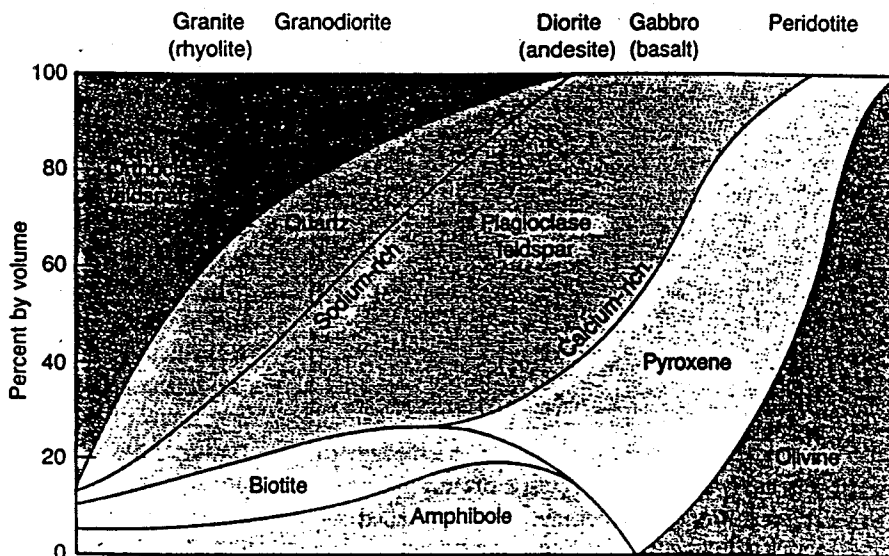


FIGURE 5.18 Classification of sedimentary rocks: inorganic detrital.

Composition	Grain-size Class and Diameter	Comments	Name		
Mainly quartz, feldspar, rock fragments, and clay minerals	gravel (> 2mm)	rounded grains	CONGLOMERATE		
		angular grains	BRECCIA		
	sand (0.0625–2.00, or 1/16–2, mm)	mostly quartz grains	QUARTZ SANDSTONE	SANDSTONE	
		mostly feldspar grains	ARKOSE		
		mostly rock fragments	LITHIC SANDSTONE		
		mixed with much silt and clay	WACKE		
	mud	silt (0.0039–0.0625, or 1/256–1/16, mm)	nonfissile (compact)	SILTSTONE	MUDSTONE
			fissile (splits easily)	SHALE	
		clay (< 0.0039, or 1/256, mm)	nonfissile (compact)	CLAYSTONE	
			fissile (splits easily)	SHALE	

Composition	Comments	Grain-size	Name	
Mainly calcium carbonate, CaCO ₃	shells or shell fragments (i.e., skeletal grains) well cemented to form dense rock	gravel (> 2 mm)	CALCIUDITE	SKELETAL LIMESTONE
		sand (0.0625–2 mm)	CALCARENITE	
		silt (0.0039–0.0625 mm)	CALCISILTITE	
		clay (< 0.0039 mm)	MICRITE	
	shells or shell fragments (i.e., skeletal grains) poorly cemented to form porous, earthy rock	gravel (> 2 mm)	COQUINA	SKELETAL LIMESTONE
		sand (0.0625–2 mm)	CALCARENITE	
		silt and clay (< 0.0625 mm)	CHALK	
	spherical grains with concentric laminations	< 2 mm	OOLITIC LIMESTONE	SKELETAL LIMESTONE
	crystals formed as inorganic chemical precipitates	coarse-grained (> 2 mm) to fine-grained (0.0039 mm)	CRYSTALLINE LIMESTONE	
		very fine grained (< 0.0039 mm)	MICRITE	
Mainly dolomite, CaMg(CO ₃) ₂	commonly altered from limestone	all sizes	DOLOSTONE	
Mainly varieties of quartz, SiO ₂ (chalcedony, flint, chert, opal, jasper, etc.)	layers, lenses, nodules	microcrystalline or amorphous	CHERT	
Mainly halite, NaCl	crystals formed as inorganic chemical precipitates	all sizes	ROCK SALT	
Mainly gypsum, CaSO ₄ · 2H ₂ O	crystals formed as inorganic chemical precipitates	all sizes	ROCK GYPSUM	
Mainly plant fragments	brown and porous	all sizes	PEAT	
	black and nonporous	all sizes or dense with conchoidal fracture	BITUMINOUS COAL	

FIGURE 5.19 Classification of sedimentary rocks: chemical and biochemical.

Table 12.10
**CLASSIFICATION OF
 TERRIGENOUS
 ROCKS*** (COMPARE
 WITH FIG. 12.18 FOR
 SANDSTONE
 CLASSIFICATION)

		COMPOSITION				
		Lithic fragments (e.g., chert, limestone, volcanic, granite)	Quartz	Feldspar		
GRAIN SIZE	Cobble	Cobble conglomerate (or breccia) (e.g., granite cobble conglomerate)		Quartz cobble conglomerate (or breccia)		
	Pebble	Pebble conglomerate (or breccia) (e.g., chert pebble conglomerate)		Quartz pebble conglomerate (or breccia)		
	Granule	Granule conglomerate (or breccia) (e.g., limestone granule conglomerate)		Quartz granule conglomerate (or breccia)	Feldspar granule conglomerate (or breccia)	
	Sand	Sandstone	Wacke (>10% matrix)	Lithic ¹ wacke	Quartz wacke	Feldspathic ² wacke
			Arenite (<10% matrix)	Lithic arenite	Quartz arenite	Feldspathic ² arenite
	Silt	<i>(composition cannot be evaluated because of fine grain size)</i>				
Siltstone						
Mud	Mudstone—lacking fissility Shale—showing fissility					

¹Dark, highly indurated lithic wackes are also referred to as greywackes.

²Red- or pink-colored feldspathic wackes and arenites can be referred to as arkoses.

*Modified after L. J. Suttner and J. Meyers, 1980, Field study of the petrology of sedimentary rocks in *Manual for Geologic Field Study of Northern Rocky Mountains*, Indiana University.

Table 12.6
**CHEMICAL
 COMPOSITION OF
 AVERAGE IGNEOUS
 ROCK COMPARED
 WITH AVERAGE
 COMPOSITIONS OF
 SOME SEDIMENTARY
 ROCK TYPES***

	Average Continental Igneous Rock	Average Sandstone	Average Shale	Average Limestone
SiO ₂	59.14	78.33	58.10	5.19
TiO ₂	1.05	0.25	0.65	0.06
Al ₂ O ₃	15.34	4.77	15.40	0.81
Fe ₂ O ₃	3.08	1.07	4.02	0.54
FeO	3.80	0.30	2.45	—
MgO	3.49	1.16	2.44	7.89
CaO	5.08	5.50	3.11	42.57
Na ₂ O	3.84	0.45	1.30	0.05
K ₂ O	3.13	1.31	3.24	0.33
H ₂ O	1.15	1.63	5.00	0.77
P ₂ O ₅	0.30	0.08	0.17	0.04
CO ₂	0.10	5.03	2.63	41.54
SO ₃	—	0.07	0.64	0.05
C(elemental)	—	—	0.80	—
Total	99.50	99.95	99.95	99.84

* After F. W. Clarke, 1924, Data of Geochemistry, U.S. Geol. Surv. Bul. 770.

TABLE 6. CLASSIFICATION OF METAMORPHIC ROCKS
A. FOLIATED

TEXTURE		COMPOSITION					ROCK NAME		
ORIENTED GRAINS	NON-LAYERED	Very Fine Grained	CHLORITE	MICA	QUARTZ	FELDSPAR	AMPHIBOLE	PYROXENE	SLATE
		Fine Grained							PHYLLITE
		Coarse Grained							SCHIST
	LAYERED	Coarse Grained							GNEISS

B. NON-FOLIATED

TEXTURE	COMPOSITION	ROCK NAME
Coarse Grained	Deformed Fragments of Any Rock Type	METACONGLOMERATE
Fine to Coarse Grained	Quartz	QUARTZITE
	Calcite or Dolomite	MARBLE